Physical Science Reference Tables

MOTION AND ENERGY

$\overline{v} = \frac{\Delta d}{\Delta t}$	v = velocity				
Δt	d = position				
$a = \frac{V_f - V_i}{\Lambda t}$	t = time				
	a = uniform acceleration				
F = ma	F = force				
$F_g = mg$	<i>m</i> = mass				
p = mv	F_g = weight				
$W = F \Delta d$	g = acceleration due to gravity on Earth = 9.8 m/s/s				
$P = \frac{W}{\Delta t}$	p = momentum				
Δ c	W = work				
$PE_g = mgh = F_gh$	P = power				
$KE = \frac{1}{2}mv^2$	PE_g = gravitational potential energy				
_	h = height				
$IMA = \frac{d_E}{d_R}$	KE = kinetic energy				
E	IMA = ideal mechanical advantage				
$AMA = \frac{F_R}{F_E}$	AMA = actual mechanical advantage				
Efficiency = $\frac{W_{out}}{W_{in}} \times 100$	R = resistance				
	E = effort				
$V_w = f\lambda$	v_w = wave velocity				
	<i>f</i> = frequency				
	λ = wavelength				

ELECTRICITY

V = IR	V = electrical potential difference			
P = VI	I = current			
	R = resistance			
	P = power			
DENSITY				
$D = \frac{m}{V}$	D = density			
	m = mass			
	V = volume			

PERIODIC TABLE

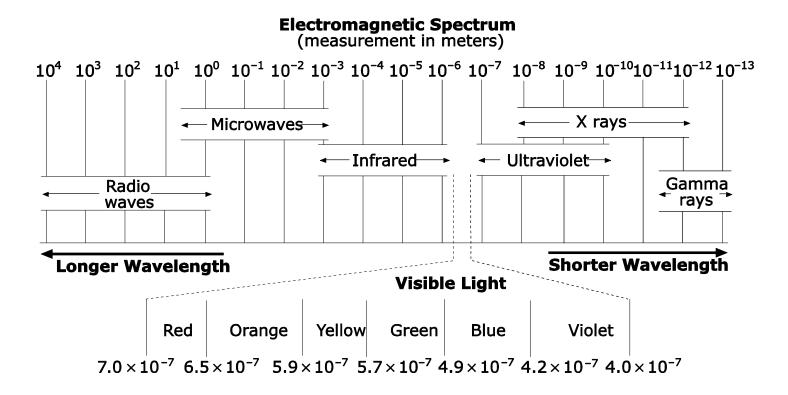
1 IA								
1 H Hydrogen 1.008	2 IIA							
3 Li Lithium 6.941	4 Be Beryllium 9.012							
11 Na Sodium 22.99	12 Mg Magnesium 24.31	3 IIIB	4 IVB	5 VB	6 VIB	7 VIIB	8 VIIIB	9 VIIIB
19	20	21	22	23	24	25	26	27
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co
Potassium	^{Calcium}	Scandium	^{Titanium}	Vanadium	^{Chromium}	^{Manganese}	^{Iron}	^{Cobalt}
39.10	40.08	44.96	47.88	50.94	51.99	54.94	55.85	58.93
37	38	39	40	41	42	43	44	45
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh
Rubidium	Strontium	^{Yttrium}	Zirconium	Niobium	^{Molybdenum}	Technetium	Ruthenium	Rhodium
85.47	87.62	88.91	91.22	92.91	95.94	(98)	101.07	102.91
55	56	57	72	73	74	75	76	77
Cs	Ba	La	Hf	Ta	W	Re	Os	Ir
Cesium	^{Barium}	Lanthanum	^{Hafnium}	^{Tantalum}	^{Tungsten}	Rhenium	^{Osmium}	Iridium
132.91	137.38	138.91	178.49	180.95	183.84	186.21	190.23	192.22
87	88	89	104	105	106	107	108	109
Fr	Ra	Ac	Rf	Db	Sg	Bh	Hs	Mt
Francium	^{Radium}	^{Actinium}	Rutherfordium	Dubnium	Seaborgium	^{Bohrium}	_{Hassium}	Meitnerium
(223)	(226)	(227)	(261)	(262)	(263)	(264)	(269)	(268)

58	59	60	61	62	63	64
Ce	Pr	Nd	Pm	Sm	Eu	Gd
^{Cerium}	Praseodymium	Neodymium	Promethium	^{Samarium}	^{Europium}	^{Gadolinium}
140.12	140.91	144.24	(145)	150.36	151.96	157.25
90	91	92	93	94	95	96
Th	Pa	U	Np	Pu	Am	Cm
Thorium	Protactinium	^{Uranium}	Neptunium	Plutonium	^{Americium}	_{Curium}
232.04	231.04	238.04	(237)	(244)	(243)	(247)

OF THE ELEMENTS

								18 VIIIA
			13 IIIA	14 IVA	15 VA	16 VIA	17 VIIA	2 He ^{Helium} 4.003
			5 B Boron 10.81	6 C Carbon 12.01	7 N Nitrogen 14.01	8 0 0xygen 16.00	9 F Fluorine 19.00	10 Ne _{Neon} 20.18
10 VIIIB	11 IB	12 IIB	13 Al Aluminum 26.98	14 Si Silicon 28.09	15 P Phosphorus 30.97	16 S ^{Sulfur} 32.07	17 Cl Chlorine 35.45	18 Ar ^{Argon} 39.95
28 Ni ^{Nickel} 58.69	29 Cu ^{Copper} 63.55	30 Zn 2inc 65.39	31 Ga ^{Gallium} 69.72	32 Ge ^{Germanium} 72.61	33 As ^{Arsenic} 74.92	34 Se ^{Selenium} 78.96	35 Br ^{Bromine} 79.90	36 Kr ^{Krypton} 83.80
46 Pd Palladium 106.42	47 Ag ^{Silver} 107.87	48 Cd ^{Cadmium} 112.41	49 In 114.82	50 Sn ^{Tin} 118.71	51 Sb Antimony 121.76	52 Te Tellurium 127.60	53 I 126.90	54 Xe Xenon 131.29
78 Pt Platinum 195.08	79 Au _{Gold} 196.97	80 Hg Mercury 200.59	81 TI Thallium 204.38	82 Pb Lead 207.20	83 Bi ^{Bismuth} 208.98	84 Po Polonium (209)	85 At Astatine (210)	86 Rn ^{Radon} (222)
110 Ds Darmstadtium (271)	111 Rg Roentgenium (272)	112 Cn Copernicium (285)						

65	66	67	68	69	70	71
Tb	Dy	Ho	Er	Tm	Yb	Lu
^{Terbium}	Dysprosium	^{Holmium}	^{Erbium}	Thulium	Ytterbium	^{Lutetium}
158.93	162.50	164.93	167.26	168.93	173.04	174.97
97 Bk Berkelium (247)	98 Cf (251)	99 Es Einsteinium (252)	100 Fm Fermium (257)	101 Md Mendelevium (258)	102 No Nobelium (254)	103 Lr Lawrencium (262)



Polyatomic Ions			
NH_4^+	Ammonium		
$C_2H_3O_2^-$	Acetate		
	Chlorate		
NO ₃	Nitrate		
OH⁻	Hydroxide		
CO ₃ ²⁻	Carbonate		
SO ₄ ²⁻	Sulfate		
PO ₄ ³⁻	Phosphate		